

WHEN PRODUCING E-466 ON THE BASIS OF LOCAL RAW MATERIALS - A COMPOSITE OF SILVER IONS GRADES APPLICATION IN SEVERAL INDUSTRIAL NETWORKS OF HOUSEHOLD CHEMICALS

Sandybaeva Zamira Hudayberdievna Osh State University

Murodov Muzaffar Murodovich, Tashkent Innovative Chemical-Technological Scientific-Research Institute

Abstract

In particular, this research was conducted on the simplification of the composition of laundry detergents as harmless as possible, at least partially free from various carcinogenic effects, as well as the exchange of local reagents, achieving environmental purity - in-depth analysis.modified during the synthesis of E-466, obtained on the basis of local raw materials - composite silver ionic brands, applied in several existing branches of household chemistry, research has been conducted and positive results have been achieved.

Keywords: pentosan, alkali sediment, suffocation, ash content, moisture, cellulose, concentration, parameter, optimal conditions, destruction, Banana cellulose, extraction process, basic substance content, cotton lint, polymerization rate.

Additives improve the quality of detergents and make it possible to drastically reduce the content of the active component in a commercial product without reducing its washing ability. The most common detergent additives are condensed phosphates, silicates, sulfates and carboxymethyl cellulose salts. Of the inorganic salts, sodium sulfate is widely used as an additive in delicate laundry detergents and dishwashing compositions. In the production of synthetic detergents in 1968, 16% of the produced sodium sulfate was used. Of the condensed phosphates, sodium tetrapyrophosphate and sodium tripolyphosphate are especially effective as additives. These salts were first used as fillers in the USA in the early 1940s. Sodium tripolyphosphate takes the first place among them in terms of consumption in the production of detergents. In 1970, the production of sodium tripolyphosphate amounted to 1079.6 thousand tons (230). In 1971, it was produced by six firms at factories with a total capacity of Mo2 thousand tons. Up to 90% of the produced sodium tripolyphosphate is used as a filler in the production of synthetic detergents.



Modern synthetic detergents are complex multicomponent mixtures. The main component is CMC (CMC), which reduces the surface tension of water, improves the wettability of the fabric, increases the emulsifying and foaming ability of detergents. The widespread use of carboxymethyl cellulose in the production of CMC (CMC) is due to its anti-resorption effect, the ability to enhance the suspending effect of inorganic electrolytes and polyphosphates, stabilize the foam and reduce the ash content of cotton fabric products.

A small addition of CMC (CMC) to washing powder or detergent paste prevents dirt particles from returning to the surface of the fabric during washing, ensuring a high degree of cleanliness.

Application:

- •cleaners
- •soap
- washing powders

Processes of studying the composition of composite compounds in household detergents and the study of cellulose esters in this composition. A quality detergent is a guarantee of cleanliness. The detergent should make the smell and quality of the fabric being washed as desired. Due to the negligence of the user, the washing machine can sometimes affect the health of the washer. The choice depending on the type of washing powder is not yet formed in most cases. Universal detergents can be used to wash any type of fabric and wool products, both in the washing machine and by hand. They also remove various dirt, stains and small stains on the fabric [1-12]. That is why the name "universal" is indicated on the powder container, which is one step higher than other types. Laundry detergent with the name "bio" on the label is considered a little stronger and helps to remove stains from sauces, coffee, tea, fruits and grease. "Automatic" - has a special bleach, but this type, which has a lower foaming property than other types, is mainly designed for washing machines [13-30]. It is better not to wash brightly colored fine silk fabrics and cotton fabrics in such a powder. Synthetic and thick fabrics can be washed in powders recommended for washing in cold water. If you want to wash in the washing machine, you do not need to heat the water. The packaging of such detergents will state that the water temperature should not exceed 60 degrees. Bio supplements are enzymes that are compatible with nature. Enzymes, in turn, have the ability to disperse various stains and eliminate complications caused by food and drink. Bio additives are exposed to 60 degrees, decomposing at higher temperatures. With the addition of powder and bio, you can wash everything except natural silk, wool, as well as fabrics made from other natural fibers. The more complex the



composition of laundry detergent, the more difficult it is to know its effects on human health. In most cases, the harmful chemicals in the engine remain on the surface of the container, enter the human body with food and cause a number of serious diseases. Dishwashing detergents are called surfactants in the language of chemists, they are divided into cationic, anionic, amphoteric, neonogenic larvae. These chemical compounds are much harder to wash off from the surface of the dish.

When the dish is washed with washing gel, water droplets contain the active ingredients of the dissolved surface. When the water dries, these substances cover the surface of the container. When hot food is placed in a bowl, the active ingredients of the surface are added to our food and enter the body along with the food.

Most of the world's population suffers from various allergic diseases. Many explain this by the deterioration of the environmental situation, 70 percent of whom do not know that allergies and other diseases are the result of the use of chemical dishwashing detergents in the kitchen. Often the onset of the disease manifests itself as an allergic reaction, ending with cancer of the gastrointestinal tract, hypertension, depression.

Of the chemicals listed above, chlorine is the safest for nature. It is broken down into components that are safe in nature. Chlorine and its organic compounds adversely affect humans, causing cardiovascular disease, atherosclerosis, anemia, hypertension. The active ingredients of the surface are carcinogenic [31-38]. They absorb the protective layer of the stomach. The compounds are absorbed into the bloodstream and gradually poison the whole organism. Not only active ingredients, but also fragrances and preservatives that are part of the dishwashing detergent have a negative effect on the human body. Wastewater, which contains dishwashing detergent, also causes great harm to the environment. When wastewater enters a body of water, it causes them to become swampy, killing animals and plants. If the dishwashing detergent contains anionic surfactants, they are almost indestructible and cause three times more damage to the human body and the environment than others. Today, a number of foreign countries have strict requirements for chemical dishwashing detergents. For example, in the European Union, in accordance with EU Regulation No 259/2012, the labels of dishwashing detergents must contain full information about the product and its composition. In addition, a few ingredients should be listed regardless of their concentration. For example, fragrances. The provisions of the Technical Regulation adopted in Russia in 2010 do not require a complete list of detergent ingredients. Manufacturers who use it effectively often "forget" the most toxic and toxic substances. In the United States, the sale of soaps and bath gels, along with chemical detergents labeled "antibacterial", is on the verge

of being banned. Many years of research have found 19 components in their composition that are dangerous to human health. The most important of these are triclocarbons and triclosan, which have so far been thought to kill viral and other disease-causing microorganisms. The ability of antimicrobial triclosan to cause liver cirrhosis and fibrosis has recently been established. According to Janet Woodock, head of the Expert Center, the negative effects of triclosan on human health have been fully established. First and foremost, the detergent must be safe for human health. The above are the current events and the detergents are as harmless as possible, taking into account the negative consequences arising from them. This research was carried out on the simplification of the composition, albeit partially free from various carcinogenic effects, as well as the exchange through local reagents, to achieve environmental purity - in-depth analysis. The following is a composite composition of washing powder obtained on the basis of local raw materials, according to which the composition of the composite has a high degree of exchange of the drug E-466 and a brand with a high base content and high viscosity was used:

Consumption rate of reagents for laundry detergent based on the ingredients available and recommended in production TABLE-1

Nº	Chemical reagents	Available content, consumption of reagents - norm, kg	Consumption of the proposed content - norm, kg
1	$\mathrm{Na_{2}CO_{3}}$	20,2	28,8
2	$\mathrm{Na_{2}SO_{4}}$	12,2	8,4
3	$\mathrm{Na_{2}PO_{4}}$	4,6	4,2
4	LAS-80	7,25	5,4
5	$\mathrm{Na_{2}SiO_{3}}$	1,8	-
6	NaCl	0,5	-
7	Na-KMTs	0,450	0,280 (E-466)
8	$\mathrm{H_{2}O}$	2,6	2,4
9	Glycerin	1,4	0,5
	Total	50,0	50,0

It can be seen from the table that due to the reduction of sharp consumption norms of the existing content in production, as well as the addition of an innovative product - E-466 to the new composition, under its influence, the clothes being washed are transparent and it became clear that I was achieving a positive result in mussafo. That is, aqueous solutions of simple esters of cellulose (Na-KMTs, E-466) are able to significantly increase the viscosity (viscosity) of latexes and various oils.

When Na-KMTs, a simple ester of up to 25% cellulose, are added to E-466, the cleaning effect of the corresponding detergents is so improved that it is equivalent to natural soap. The inclusion of the synthesized E-466 innovative product in the composition on the basis of strictly structured consumption norms gave the above positive conclusions.

Study of silver ion E-466 as a reagent stabilizing the composition of washing powder and its role in the composite composition. Antibacterial, disinfectant washing powder is used both in laundry and at home. Instead of strong substances, silver ionic washing powder is a safe and very effective antibacterial detergent. The detrimental effects of silver ions, as well as its beneficial properties have long been scientifically proven, that is, the bactericidal, antiviral, antifungal, disinfectant properties of silver ions are incomparable.

Even small concentrations of silver ions in washing powder are able to destroy various harmful cell membranes.

The advantages of silver ion detergent are as follows the disinfectant effect of silver is 3.5 times higher than that of sodium hydrochloride, silver does not form toxic compounds and has no odor. Silver ions affect more than 350 species of bacteria.

A study of the use of modified silver ion E-466 brands in the composition of laundry detergent in the fight against various bacteria and viruses and the proposal of positive results.

Taking into account the above, during the dissertation research it was recommended to produce an innovative composite product with a new content of washing powder consumption - the norm. The difference between this composition and existing analogues is that the silver ion was initially synthesized on the basis of local raw materials, i.e. as a result of the interaction of silver fragments with the mineral acid - HNO₃. The silver ion in the resulting acid medium was neutralized and dried using special filters and incorporated into the composite during the synthesis of Na-KMTs. Resistant to silver acid. Chloride, dilute sulfuric acid and "aqua regia" do not affect this, because on the surface of the metal formed a protective film of silver chloride (AgCl). Silver dissolves well in nitric acid and soluble sodium nitrate (AgNO₃) is formed:

$Ag + 2HNO_3 = AgNO_3 + NO_2 + H_2O$

Hot concentrated sulfate acid of silver dissolves to form silver sulfate (Ag2SO4). Silver sulfate has a water solubility of 20 $^{\circ}$ C and a weight of 0.79%.

A clear bactericidal effect of silver ions above 150 micro g/l, i.e. the ability to clearly kill certain bacteria, is observed. At a concentration of 50-100 microg / l, silver ions have a bactericidal effect i.e. the ability to block the growth and proliferation of



bacteria. It is important to note that Bacteriostasis is a reversible process and the influencing factor can grow and multiply once it is over. Only cases of long-term bacteriostatic effect are excluded. The main conclusion is that silver in 50 microg/l of water has the best bacteriostatic effect at the concentrations allowed by the current standards - SanPin, i.e. it can dramatically slow down bacterial growth.

In the dissertation research, the consumption of E-466 and silver ion, which are included in the composition of laundry detergent with a new consumption rate, was determined as a result of many repeated studies and recommended for production. Addition of Na-KMTs in the binding of silver ions in the localization process following the mercerization process, as well as the addition of various reagents in the exchange process and the direct involvement of cellulose with functional groups at the level of possible exchange in hydroxy groups and was used as a key raw material in the synthesis of several brands of E-466, an innovative product as a result of extraction using ethyl alcohol - an organic compound with an important composite content and several types of products based on it.

In conclusion, it should be noted that E-466 modified during the synthesis of E-466 obtained on the basis of local raw materials - composite silver ion brands, has been studied for application in several existing branches of household chemistry and achieved positive results. In particular, this research was conducted on the simplification of the composition of laundry detergents as harmless as possible, at least partially free from various carcinogenic effects, as well as the exchange of local reagents, achieving environmental purity - in-depth analysis. The following is a composite composition of washing powder obtained on the basis of local raw materials, according to which the composition of the composite has a high degree of exchangeability of the drug E-466 and a brand with a high base substance content and high viscosity was used. Available in production on this basis and reagents for laundry detergent based on the proposed composition were developed and submitted for introduction separately for consumption.

In this chapter, the study of silver ion E-466 as a reagent for stabilizing the composition of laundry powder and its role in the composite composition was carried out, and significant positive results were obtained by incorporating silver ions into the synthesized drug E-466 on the basis of composite-modification. It is known that antibacterial, disinfectant washing powder is used both in laundry and at home. Instead of strong substances, silver ionic washing powder is a safe and very effective antibacterial detergent. The detoxifying effects of silver ions, as well as its beneficial properties have long been scientifically proven, i.e. the bactericidal, antiviral, antifungal, disinfectant properties of silver ions are incomparable.

It should be noted that Bacteriostasis is a reversible process and after the end of the influencing factor, growth and reproduction are possible. Only cases of long-term bacteriostatic effect are excluded. The main conclusion is that silver in 50 microg / l of water has the best bacteriostatic effect at permissible concentrations - SanPin, according to current standards, i.e. it can drastically slow down bacterial growth.

In the dissertation research, the consumption of E-466 and silver ion, which are included in the composition of laundry detergent with a new consumption rate, was determined as a result of many repeated studies and recommended for production. The addition of the Na-KMTs synthesis period in the binding of the silver ion to the consumption rate determined in the composite during the localization process following the mercerization process and the direct involvement of cellulose with various reagents and functional groups in the exchange process at the level of possible exchange in hydroxide groups, as well as in the synthesis of several brands of E-466, an innovative product by ethyl alcohol extraction - an important composite organic substance and was used as the main raw material in the synthesis of several types of products based on it.

REFERENCES

- [1] M.M. Murodov. «Technology of making cellulose and its ethers by using raw materials» // International Conference "Renewable Wood and Plant Resources: Chemistry, Technology, Pharmacology, and Medicine". Saint-Petersburg, Russia. June 21-24., 2011. 142-143.
- [2]. M.M. Murodov. «The technology of making carboxymethyl cellulose (cmc) by method monoapparatus» // International Conference «Renewable Wood and Plant Resources: Chemistry, Technology, Pharmacology, and Medicine». Saint-Petersburg, Russia. June 21-24., 2011. 141-142.
 - [3]. Ўзбекистон Республика Вазирлар Маҳкамаси "РЕСПУБЛИКАДА ТЕЗ ЎСУВЧИ ВА САНОАТБОП ПАВЛОВНИЯ ДАРАХТИ ПЛАНТАЦИЯЛАРИНИ БАРПО ҚИЛИШ ЧОРА-ТАДБИРЛАРИ ТЎҒРИСИДА" 2020 йил 27 августдаги 520-сонли қарори.
 - [4]. Интернет: https://xs.uz/uzkr/post/ hududlarda –pavlovniya -plantatsiyalari -tashkil-qilinadi/
 - [5]. Муродов, М. Х., & Муродов, Б. Х. У. (2015). Фотоэлектрическая станция с автоматическим управлением мощностью 20 кВт для учебного заведения. Science Time, (12 (24)), 543-547.



- [6]. Murodov, M. M., Rahmanberdiev, G. R., Khalikov, M. M., Egamberdiev, E. A., Negmatova, K. C., Saidov, M. M., & Mahmudova, N. (2012, July). Endurance of high molecular weight carboxymethyl cellulose in corrosive environments. In AIP Conference Proceedings (Vol. 1459, No. 1, pp. 309-311). American Institute of Physics.
- [7]. Murodov, M. M., Yusupova, N. F., Urabjanova, S. I., Turdibaeva, N., & Siddikov, M. A. (2021). OBTAINING A PAC FROM THE CELLULOSE OF PLANTS OF SUNFLOWER, SAFFLOWER AND WASTE FROM THE TEXTILE INDUSTRY.
- [8]. Murodov, M. M., Yusupova, N. F., Urabjanova, S. I., Turdibaeva, N., & Siddikov, M. A. Obtaining a Pac From the Cellulose of Plants of Sunflower, Safflower and Waste From the Textile Industry. European Journal of Humanities and Educational Advancements, 2(1), 13-15.
- [9]. Murodov, M. M., Xudoyarov, O. F., & Urozov, M. Q. (2018). Technology of making carboxymethylcellulose by using local raw materials. Advanced Engineering Forum Vols. 8-9 (2018) pp 411-412/©. Trans Tech Publications, Switzerland. doi, 10, 8-9.
- [10]. Primqulov, M. T., Rahmonbtrdiev, G., Murodov, M. M., & Mirataev, A. A. (2014). Tarkibida sellyuloza saqlovchi xom ashyoni qayta ishlash texnologiyasi. Ozbekiston faylasuflar milliy jamiyati nashriyati. Toshkent, 28-29. [11]. Рахманбердиев, Г. Р., & Муродов, М. М. (2011). Разработка технологии получения целлюлозы из растений топинамбура. Итисодиёт ва инновацион технологиялар" илмий электрон журнали,(2), 1-11.
- [12]. Elievich, C. L., Khasanovich, Y. S., & Murodovich, M. M. (2021). TECHNOLOGY FOR THE PRODUCTION OF PAPER COMPOSITES FOR DIFFERENT AREAS FROM FIBER WASTE.
- [13]. MURODOVICH, M. M., QULTURAEVICH, U. M., & MAHAMEDJANOVA, D. (2018). Development of Technology for Production of Cellulose From Plants of Tissue and Receiving Na-Carboxymethylcellulose On its Basis. JournalNX, 6(12), 407-411.
- [14]. Rahmonberdiev, G., Murodov, M., Negmatova, K., Negmatov, S., & Lysenko, A. (2012). Effective Technology of Obtaining The Carboxymethyl Cellulose From Annual Plants. In Advanced Materials Research (Vol. 413, pp. 541-543). Trans Tech Publications Ltd.
- [15]. Murodovich, M. M., Murodovich, H. M., & Qulturaevich, U. M. (2020). Obtaining technical carboxymethyl cellulose increased in main substance. ACADEMICIA: AN INTERNATIONAL MULTIDISCIPLINARY RESEARCH JOURNAL, 10(12), 717-719.



- [16]. Murodovich, M. M., Qulturaevich, U. M., & Mahamedjanova, D. Comparative Researches of the Composition and Properties Cmc in Different Degree of Polymerization. JournalNX, 6(12), 412-415.
- [17] Йулдашева, Г. И., & Тешабаева, О. Н. (2020). Развитие цифровой экономики Республики Узбекистан. Universum: экономика и юриспруденция, (7 (72)), 4-6. [18] Teshabaeva, О., Yuldasheva, G., & Yuldasheva, M. (2021). DEVELOPMENT OF ELECTRONIC BUSINESS IN THE REPUBLIC OF UZBEKISTAN. Интернаука, (3-3), 16-18.
- [19] Ibragimovna, Y. G. (2022). ADVANTAGES OF CREDIT-MODULE SYSTEM IN THE FIELD OF EDUCATION. INTERNATIONAL JOURNAL OF SOCIAL SCIENCE & INTERDISCIPLINARY RESEARCH ISSN: 2277-3630 Impact factor: 7.429, 11, 14-16.
- [20] Йўлдашева, М. (2021). ЭФФЕКТИВНОЕ УПРАВЛЕНИЕ ИНВЕСТИЦИОННОЙ ДЕЯТЕЛЬНОСТЬЮ ИНФОРМАЦИОННО-КОММУНИКАЦИОННЫХ ТЕХНОЛОГИЙ УЗБЕКИСТАНА. Студенческий вестник, (3-4), 11-13.
- [21] Shermatova, G. Y. H. (2022). ANIQ FANLARNI O'QITISHDA AXBOROT TEXNOLOGIYALARIDAN FOYDALANISH. Scientific progress, 3(1), 372-376.
- [22] Yuldasheva, G. I., & Shermatova, K. M. (2021). THE USE OF ADAPTIVE TECHNOLOGIES IN THE EDUCATIONAL PROCESS. Экономика и социум, (4-1), 466-468.
- [23] Худаёрова, С. И. (2022). ОСОБЕННОСТИ МОРФОЛОГИЧЕСКОГО ФОРМИРОВАНИЯ ЛИСТЬЕВ У СОРТОВ ЛИМОНА (CITRUS L.) В ЗАЩИЩЕННЫХ МЕСТАХ. БАРҚАРОРЛИК ВА ЕТАКЧИ ТАДҚИҚОТЛАР ОНЛАЙН ИЛМИЙ ЖУРНАЛИ, 15-18.
- [24] Қодирова, Г. О. Қ., & Худоёрова, Ф. (2021). РОЛЬ ОБРАЗОВАТЕЛЬНЫХ ТЕХНОЛОГИЙ В ПРЕПОДАВАНИИ ЯЗЫКА. Scientific progress, 2(3), 894-898.
- [25] Itolmasovna, K. S. (2022). DEVELOPMENT OF MARKETABLE PROPERTIES OF PROCESSED LEMON. The American Journal of Agriculture and Biomedical Engineering, 4(02), 21-25.
- [26] Муродов, М. М., & Чулиев, Л. Э. (2021, October). Турли Объектлар Асосида, Яъни Пахта Тозалаш Корхоналарининг Толали Чикиндилари Ва Павлония Хамда Банан Целлюлозаларидан Е-466 Олиш Технологияси Ва Унинг Физик-Кимёвий, Механик-Структуравий Хоссалари. In "ONLINE-CONFERENCES" PLATFORM (pp. 316-320).



- [27] Ibragimkhodjayev, A. M., Rakhmonberdiyev, G. R., Murodov, M. M., & Kodirov, O. S. (2009). "Influence of ripening process of cellulose from topinambour on its fractional composition. Chemistry and chemical technology. Tashkent, (4), 57.
- [28] Муродов, М. М., & Чулиев, Л. Э. (2021, October). Махаллий Хом Ашёлар Асосида Олинган Целлюлозалардан, Фармацефтика Ва Медецина Сохалари Учун Юқори Тозаликка Эга Е-466 Нинг Бир Нечта Маркаларини Олиш Технологияси. In "ONLINE-CONFERENCES" PLATFORM (pp. 309-315).
- [29] Муродов, М. М. (2021). ПАВЛОНИЯ ДАРАХТИ ХАМДА БАНАН ПОЯСИ АСОСИДАГИ ЦЕЛЛЮЛОЗА СИНТЕЗИ ЖАРАЁНЛАРИНИ ЎРГАНИШ ВА СОЛИШТИРМА ХАРАКТЕРИСТИКАСИНИ ТУРЛИ БОСҚИЧЛАР ЁРДАМИДА ТАҲЛИЛ ҚИЛИШ. Scientific progress, 2(6), 1806-1813.
- [30] Муродов, М. М. (2021). ПАХТА ТОЗАЛАШ КОРХОНАЛАРИНИНГ ТОЛАЛИ ЧИҚИНДИЛАРИ ХАМДА ПАВЛОНИЯ ДАРАХТИ ВА БАНАН ПОЯЛАРИ АСОСИДА ОЛИНГАН ЦЕЛЛЮЛОЗАЛАРДАН, ФАРМАЦЕФТИКА ВА МЕДЕЦИНА СОХАЛАРИ УЧУН ЮҚОРИ ТОЗАЛИККА ЭГА Е-466 НИНГ БИР НЕЧТА МАРКАЛАРИНИ ОЛИШ ТЕХНОЛОГИЯСИ. Scientific progress, 2(6), 1814-1823.
- [31] Муродов, М. М. (2021). ТУРЛИ ОБЪЕКТЛАР АСОСИДА, ЯЪНИ ПТКТЧ ВА ПАВЛОНИЯ ХАМДА БАНАН ЦЕЛЛЮЛОЗАЛАРИДАН Е-466 ОЛИШ ТЕХНОЛОГИЯСИ ВА УНИНГ ФИЗИК-КИМЁВИЙ, МЕХАНИК-СТРУКТУРАВИЙ ХОССАЛАРИ. Scientific progress, 2(6), 1824-1831.
- [32] Муродов, М. М. (2021). МАХАЛЛИЙ ХОМАШЁЛАР АСОСИДА ОЛИНГАН Е-466 СИНТЕЗИ ДАВРИДА МОДЕФИКАЦИЯЛАНГАН—КОМПОЗИТ КУМУШ ИОНЛИ МАРКАЛАРИДАН, МАЙИШИЙ КИМЁНИНГ МАВЖУД БИР НЕЧТА ТАРМОҚЛАРИДА ҚЎЛЛАШ. Scientific progress, 2(6), 1782-1791.
- [33] Муродов, М., & Матчанова, Ф. (2020). СОЛНЕЧНАЯ БИОЭНЕРГЕТИЧЕСКАЯ УСТАНОВКА ДЛЯ УТИЛИЗАЦИИ БЫТОВЫХ ОТХОДОВ. InterConf.
- [34] Муродов, М. М., Сидиков, А. С., & Уразов, М. К. (2019, September). 4.11. ПОЛУЧЕНИЕ Na-КМЦ И ПАЦ ИЗ ЦЕЛЛЮЛОЗЫ РАСТЕНИЙ ПОДСОЛНЕЧНИКА, САФЛОРА И ИЗ ОТХОДОВ ТЕКСТИЛЬНОЙ ПРОМЫШЛЕННОСТИ. In VI Международная конференция (р. 237).
- [35] Муродов, М. М., Урозов, М. К., & Улуков, Ж. Т. САНОАТ КОРХОНАЛАРИНИНГ ТОЛАЛИ ЧИҚИНДИЛАРИНИ КИМЁВИЙ ҚАЙТА ИШЛАШ.
- [36] Муродов, М. М., & Эшонкулов, М. Н. КАВРАК ЎСИМЛИГИ ЦЕЛЛЮЛОЗАСИ АСОСИДА Na-КМЦ СИНТЕЗИ ДАВРИДАГИ ОПТИМАЛ ПАРАМЕТРЛАРНИ АВТОМАТЛАШТИРИШ БОСКИЧИГА ЎЗЛАШТИРИШ. In КОНФЕРЕНЦИЯ-СИМПОЗИУМ (р. 55).