



## IN CONDENSED HETEROCYCLIC COMPOUNDS KABACHNIK-FELDS REACTIONS

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### Abstract:

In the article, 2,3-trimethylene-3,4-dihydrobenzo[2,3-d] pyrimidin -4 - one was synthesized from 2-aminobenzoic acid and pyrrolidone-2 in the presence of various substances, for example, PCl<sub>5</sub>, POCl<sub>3</sub>. Its reduction reaction with NaBH<sub>4</sub> was carried out. The obtained 2,3-trimethylene-1,2,3,4-tetrahydrobenzo [2,3-d] pyrimidin – 4 - one - a three-component combination of carbonyl, amine and hydrophosphoryl compound leads to α-aminophosphonates, phosphoric acid-formaldehyde; reactions with aldehydes in the three-component system were studied.

**Keywords:** 2, 3- trimethylene-3,4-dihydrobenzo[2,3-d] pyrimidin-4 -one, 2,3-trimethylene- 1,2,3,4-tetrahydrobenzo[2,3d] pyrimidine - 4-one, 2-Aminobenzoic acid, pyrrolidone-2, Synthesis reactions.

### INTRODUCTION

In the world, the synthesis of new physiologically active derivatives of benzo[2,3-d]pyrimidine and the creation of modern drugs based on them are carried out with the help of high technologies. It is known that the anti-cancer drugs used up to now destroy dangerous cancer cells while simultaneously damaging healthy cells.

Representatives of benzo[2,3-d]pyrimidines, the preferred polycyclic anticancer drugs, pyrimid and pyrimid acids with antibacterial effects have been developed by world scientists. These medical institutions have a practical interest in the properties of this substance.

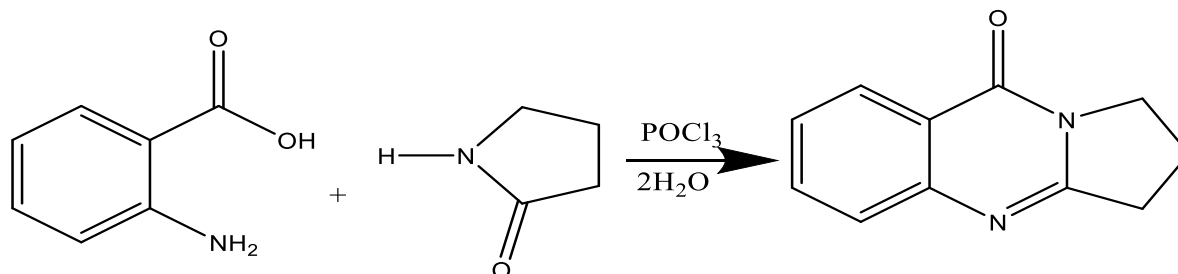
Organophosphorus compounds are ubiquitous in nature and are used in agriculture, medicine, and industry [1-3]. Some organophosphorus compounds are important pesticides [4], bactericides [5-7] and antibiotics [5]. Phosphorus analogs of α-pyrones act as HIV protease inhibitors [8]. α-Aminophosphate acids are important motifs among organophosphorus compounds in medicinal chemistry due to their clear structural similarities to α-amino acids [9-16].





## MATERIALS AND METHODS

Tricyclic 2,3-trimethylenebenzo[2,3-d]pyrimidin-4-one (1) was synthesized from 2-aminobenzoic (anthranilic) acid, which was condensed with pyrrolidone-2. The presence of  $\text{POCl}_3$ .



Aminobenzoic (anthranilic) and 2-aminopyridine-3-carboxylic acids lactam series with 2-aminophen-3 carboxylic, 2- reduce the output of the condensation product. So, if 2,3-trimethylenethieno[2,3-d]-pyridin-4-one is obtained with 81% yield, and 2,3-polymethylene-quinazolone with 56-70% yield, in 2 cases, 2,3-trimethylenebenzo-[2,3-d]pyrimidin-4-one (1) it was 74%. This fact is explained by the fact that the p-electron leakage is facilitated by the excess thiophenic ring, and the p-deficient benzene ring increases this condensation.

It is clear that the reconnection of  $\text{N}_1=\text{C}_2$  to  $\text{HN}-\text{CH}$  leads to a significant shift of the chemical shifts of protons of methylene groups towards strong fields. A significant shift of the signals of the protons of the pyridine ring after reduction was also detected. Thus, at the initial stage they appear in ppm: 7.29 (1H, dd,  $J=4.5, 7.9$ , C6-H), 8.53 (1H, dd,  $J=2.0, 7.9$ , C5-H), 8.89 (1H, dd,  $J=2.0, 4.6$ , C7-H), respectively; and in the regeneration product, the same protons are observed in the following regions: 6.74 (1H, dd,  $J=5.0; 7.6$ , C6-H), 8.06 (1H, dd,  $J=1.8; 7.4$ , C5-H) and 8.10 (1H, dd,  $J=1.8; 5.0$ , C7-H), respectively; that is, here the shift of chemical changes of protons towards strong fields. The asymmetric carbon proton of C2-H was identified in 5.10.

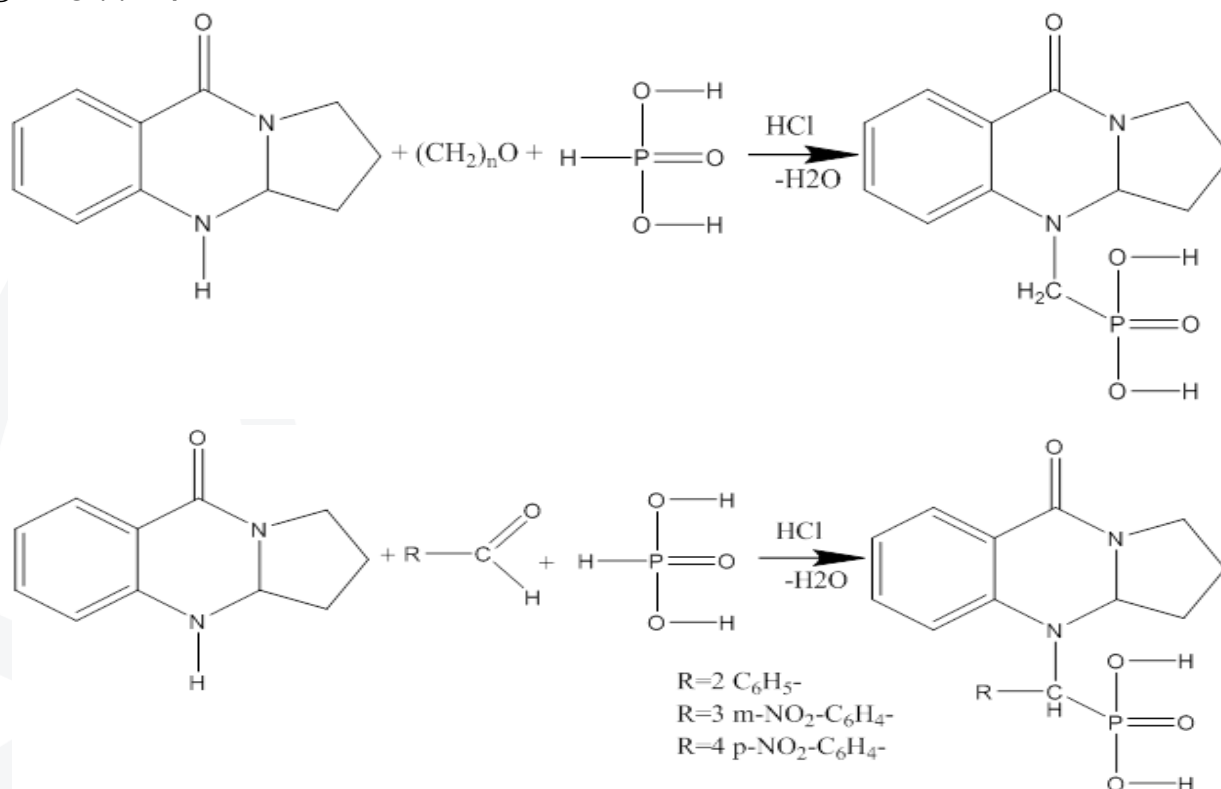
## ANALYSIS OF RESULTS

One-pot, three-component reaction of 2,3-trimethylene-1,2,3,4 with dihydrobenzo[2,3-d]pyrimidin-4-one, p and m-nitro-benzaldehyde and phosphoric acid in preliminary experiments Optimization of reaction conditions was selected as the model reaction for The procedures followed for the synthesis of  $\alpha$ -aminophosphonates in this work are shown in Table 1, entries 3–5. Product  $\alpha$ -aminophosphonates were obtained by solvent-free microwave irradiation of aldehyde, amine, and phosphoric acid for 1 min. In toluene, without any catalyst, the resulting product was well formed.



Thin layer chromatography (TLC) was used to monitor the reaction and determine the purity of the products. The reaction was carried out with a catalytic amount of HCl in toluene for 30 min. All named compounds are readily soluble in polar organic solvents.

IR spectra of compounds (3-5) showed N-CH<sub>2</sub> band at 1550 cm<sup>-1</sup>. A sharp band observed in the range 1240–1291 cm<sup>-1</sup> is caused by P=O, and the band for P-C stretching appeared in the range 740-770 cm<sup>-1</sup>. All stretching frequencies are compiled in Table 2. H NMR spectra of compounds (3-5) in 1DMSO-d<sub>6</sub> solvent were recorded. Aromatic protons of α-aminophosphonic acids appeared as a multiplet in the region δ 6.15-8.69. The proton of the P-C-H group resonated as a multiplet in the region δ 3.77–4.86.



## EXPERIMENTAL PART

Synthesis of 2,3-trimethylene-3,4-dihydrobenzo[2,3-d] pyrimidin- 4- one hydrochloride (1).

POCl<sub>3</sub> (90 mL) was added to a mixture of 2-aminobenzoic acid (65.3 g, 0.47 mol) and pyrrolidone-2 (47.94 g, 0.564 mol) and stirred at room temperature. The reaction mixture was heated at 80-90°C for 2-3 hours, cooled to room temperature and 100 ml of water was added.



The aqueous solution is treated with 5% NaOH solution to pH 7.5-8. Extracted with chloroform (3x100 ml) and the organic phase was dried with Na<sub>2</sub>SO<sub>4</sub>, filtered and the solvent evaporated. The yield of compound 1 is 30.76 g (71%), mp 139-140°C. R<sub>f</sub>=48 (10:1 chloroform-methanol): for a solution of 28.05 g (0.15 mol) 2,3-trimethylene-3,4-dihydrobenzo[2,3-d]pyrimidine-4. (1) 30 g (0.78 mol) of sodium borohydride was added to 350 ml of alcohol. The reaction mixture was boiled in a water bath for 3 hours and left for twenty-four hours. The solvent was removed, the remaining product was dissolved in water and left for 1 hour. The precipitate was filtered, washed 3 times with water, dried and recrystallized from hexane. The yield of compound 2 is 28.35 g (91%), T<sub>s</sub>= 136-139°C. R<sub>f</sub> =0.58 (10:1 chloroform-methanol).

Synthesis of compound 3. A mixture of paraform (0.005 mol), 2,3-trimethylene-1,2,3,4-tetrahydrobenzo[2,3d]pyrimidin-4-one (0.005 mol) and phosphidic acid in dry toluene was stirred for 1 hour. at room temperature min. Then the temperature was raised to 80-95 oC for 3 hours. The reaction was monitored by TLC. After completion of the reaction, toluene was removed by distillation and the residue was purified by column chromatography (5:4, benzene:hexane). The yield of compound 3 was 1.124 g (91%), T<sub>s</sub>=167-169°C. R<sub>f</sub> =0.4 (5:4, benzene:hexane)

Synthesis of compound 4. A mixture of m-nitrobenzaldehyde (0.005 mol), 2,3-trimethylene-1,2,3,4tetrahydrobenzo[2,3-d]pyrimidin-4-one (0.005 mol) and phosphidic acid was mixed in dry toluene. for 10 minutes at room temperature. The temperature was then raised to reflux for 5 h. the reaction was monitored by TLC. After completion of the reaction, toluene was removed by distillation and the residue was purified by column chromatography (5:4, benzene:hexane). The yield of compound 4 was 1.76 g (89%), T<sub>s</sub>= 155-158°C. R<sub>f</sub>=0.52 (5:4, benzene:hexane).

Synthesis of compound 5. A mixture of p-nitrobenzaldehyde (0.01 mol), 2,3-trimethylene-1,2,3,4-tetrahydrobenzo[2,3-d]pyrimidin-4-one (0.01 mol) and phosphidic acid was mixed in dry toluene. . for 10 minutes at room temperature. The temperature was then raised to reflux for 5 h. the reaction was monitored by TLC. After completion of the reaction, toluene was removed by distillation and the residue was purified by column chromatography (5:4, benzene:hexane). The yield of compound 5 was 1.78 g (90%), T<sub>s</sub>= 149-152°C. R<sub>f</sub> =0.4 (5:4, benzene:hexane)

## CONCLUSIONS

A new method for the preparation of 2,3-trimethylenebenzo[2,3-d]-pyrimidin-4-one condensation with a lactam. performs selective update of 'sh gardens.

Synthesis of a novel α-aminophosphonic acid was achieved in high yield using single-ring tri-cata.





Component reaction process, Kabachnik-Fields reaction. It involves reactions between 2, 3- trimethylene-1,2,3,4-tetrahydrobenzo[2,3-d] pyrimidin-4-one, substituted aromatic aldehydes, and phosphoric acid in dry toluene at reflux temp. Their structures were determined by IR elemental analysis, <sup>1</sup>H-NMR and mass spectrum data.

## LITERATURE

1. Azamatova M. et al. HEALING PROPERTIES OF CHAMOMILE //Академические исследования в современной науке. – 2023. – Т. 2. – №. 8. – С. 37-40.
2. Ulfat o'g'li O. A. et al. Biological Activity of the Coordinating Compound of Qinazolin-4-One with Nickel-(II) Nitrate Hexahydrate //Texas Journal of Agriculture and Biological Sciences. – 2023. – Т. 13. – С. 61-63.
3. Sapaev B., Saitkulov F. CHROMATO MASS SPECTROMETRIC ANALYSIS USING ESSENTIAL OILS //Международная конференция академических наук. – 2023. – Т. 2. – №. 1. – С. 123-126.
4. Kholmiraev M., Khaydarov G., Saitkulov F. DETERMINATION OF SIMPLE ESTERS IN MINT BY THE METHOD OF CHROMATOMASS SPECTROSCOPY //International Bulletin of Medical Sciences and Clinical Research. – 2023. – Т. 3. – №. 1. – С. 70-72.
5. Kholmiraev M., Khaydarov G., Saitkulov F. SPECTRAL CHARACTER OF THE SIMPLE OF ETHERS //Theoretical aspects in the formation of pedagogical sciences. – 2023. – Т. 2. – №. 2. – С. 204-206.
6. Tilyabov M., Khaydarov G., Saitkulov F. CHROMATO MASS SPECTROMETRIC ANALYSIS USING MINT ESSENTIAL OILS //International Bulletin of Medical Sciences and Clinical Research. – 2023. – Т. 3. – №. 1. – С. 57-60.
7. Tilyabov M., Khaydarov G., Saitkulov F. CHROMATOGRAPHY-MASS SPECTROMETRY AND ITS ANALYTICAL CAPABILITIES //Development and innovations in science. – 2023. – Т. 2. – №. 1. – С. 118-121.
8. Tilyabov M., Khaydarov G., Saitkulov F. THE STUDY OF ESTERS CHROMATOGRAPHY-MASS SPECTROMETRY OF ABSOLUTE ETHANOL EXTRACT OF THE CENTRAL ASIAN MINT PLANT (LAMIACEAE) //International Bulletin of Medical Sciences and Clinical Research. – 2023. – Т. 3. – №. 1. – С. 61-65.
9. Kholmiraev M., Khaydarov G., Saitkulov F. METHOD OF OBTAINING HERBAL MINT EXTRACT //International Bulletin of Medical Sciences and Clinical Research. – 2023. – Т. 3. – №. 1. – С. 66-69.





10. Saitkulov F. et al. STUDYING THE PROPERTIES OF THE UV SPECTRA OF QUINAZOLIN-4-ONES AND-TIONS //Development and innovations in science. – 2023. – Т. 2. – №. 1. – С. 24-27.
11. Saitkulov F. et al. STUDY OF THE STRUCTURE OF 2, 3-DIMETHYLQUINAZOLINE-4-ON //Академические исследования в современной науке. – 2023. – Т. 2. – №. 2. – С. 5-10.
12. Саиткулов Ф. Э., Элмурадов Б. Ж., Гиясов К. АЛКИЛИРОВАНИЯ ХИНАЗОЛИН-4-ОНА «МЯГКИМ» И «ЖЕСТКИМ» АЛКИЛИРУЮЩИМИ АГЕНТАМИ //Universum: химия и биология. – 2023. – №. 1-2 (103). – С. 53-56.
13. Murodillayevich K. M., Shoyimovich K. G., Ergashevich S. F. Studying the Aroma of Mint Essential Oil //INTERNATIONAL JOURNAL OF BIOLOGICAL ENGINEERING AND AGRICULTURE. – 2023. – Т. 2. – №. 1. – С. 54-56.
14. Baymuratova G. O., Nasimov K. M. THE COORDINATION COMPOUND OF COBALT CHLORIDE WITH 6-BENZYLAMINOPURINE AND AN EXPLANATION OF ITS BIOLOGII ACTIVITY //Kimyo va tibbiyot: nazariyadan amaliyotgacha. – 2022. – С. 60-62.
15. Baymuratova G. O., Nasimov K. M. THE COORDINATION COMPOUND OF COBALT CHLORIDE WITH 6-BENZYLAMINOPURINE AND AN EXPLANATION OF ITS BIOLOGII ACTIVITY //Kimyo va tibbiyot: nazariyadan amaliyotgacha. – 2022. – С. 60-62.
16. Murodovich N. H. et al. PREPARATION AND STRUCTURE OF 6-BENZYLAMINOPURINE //Galaxy International Interdisciplinary Research Journal. – 2021. – Т. 9. – №. 12. – С. 724-726.

